SiO₂ MEDIATED REACTION OF ISATIN WITH *N*-HALOSACCHARINS: A REGIOSPECIFIC PREPARATION OF 5-HALOISATINS

Soraia P.L. de Souza, Joaquim Fernando M. da Silva* and Marcio C.S. de Mattos*

Instituto de Química, Departamento de Química Orgânica Universidade Federal do Rio de Janeiro, Caixa Postal 68545 21945-970, Rio de Janeiro, Brazil

Abstract: N-halosaccharins (5 mmol) react smoothly at room temperature with isatin (5 mmol) in the presence of SiO_2 (5 g) to produce specifically the 5-halo derivative. Using this procedure, 5-chloroisatin was obtained in 48 % purified yield (72 h), 5-bromoisatin in 58 % (12 h) and 5-iodoisatin in 80 % (8 h).

Introduction

Surface mediated reactions are of great utility in organic synthesis, and in many cases the products are obtained in better yields and improved selectivities than in a homogeneous media in solution (1). N-halosuccinimides proved to be efficient reagents for the halogenation of heteroaromatics in mild homogeneous or heterogeneous conditions (2).

Isatin (1*H*-indole-2,3-dione, 1) is a versatile molecule with many applications in synthetic organic chemistry (3). Furthermore, some derivatives have important biological and pharmacological properties (3b). Several reagents (4) are described in the literature to perform the halogenation of isatin mainly in the 5-position and among them *N*-haloamides and *N*-haloimides (4a, 4b). These reagents are a suitable synthetic alternative to the highly toxic and corrosive Cl₂ and Br₂, which can also lead to other products, just as when the bromination of isatin is performed with bromine in an alcoholic media, and 5,7-dibromo-3,3-dialkoxyoxindole is then obtained (5).

We communicate here our results on the SiO₂ mediated reaction of isatin with N-halosaccharins (NXSac), reagents more electrophilic (6) but much less employed than the structural analogs N-halosuccinimides in synthetic organic chemistry

Results

N-halosaccharins react smoothly with isatin in the presence of SiO₂ to produce especificaly the 5-halo derivatives 2 in moderate to good yields (Scheme and Table 1). The reaction was performed stirring together at room temperature a suspension of SiO₂ (5 g), isatin (5 mmol) and NXSac (5 mmol) in CH₂Cl₂. After work up, the product was recrystallized from aqueous etanol.

In all the reactions the purity of the products (> 95 %) was confirmed by analytical methods (¹³C NMR and high-resolution gas chromatography). The products were characterized by analytical data and by coelution with authentic samples in a high-resolution gas chromatograph. No products derived from N-halogenation of isatin (7) or in the 7-position were detected by the analytical techniques employed (¹H and ¹³C NMR and high-resolution gas chromatography).

The same reaction performed with NCSac in the absence of SiO₂ produced 5-chloroisatin in low yield along with some unidentified products and unreacted isatin after more than 75 h.

Scheme.

2	t (h)	Yield (%) ^a
X = C1	72	48
X = Br	12	58
X = I	8	80

^a Yield of pure product based on isatin.

Table 1. Yields of 2 from the SiO₂ mediated reaction of isatin with NXSac.

In summary, we present here a convenient regiospecific preparation of 5-haloisatins, important precursors of antimicrobial compounds (8). Furthermore, the reaction conditions are mild, there is no need of special techniques or harzadous reagents and the isolation of the product is very easy.

Experimental

NCSac (9), NBSac (9) and NISac (6) were prepared as described in the literature. SiO₂ (Aldrich, 270-70 MESH, 60 Å) was activated by heating for 1.5 min in a domestic microwave oven at highest power (10). Isatin (Aldrich, Gold label) and other chemicals were used without further purification. ¹H and ¹³C NMR were acquired on a Bruker AC-200 (200 MHz and 50 MHz, respectively) spectrometer in CDCl₃ / DMSO-d₆ solutions with TMS as internal standard. IR spectra were recorded on a Perkin-Elmer 1600 FT-IR spectrometer (KBr film). Analyses by high-resolution gas chromatography were performed on a HP-5890-II gas chromatograph with FID by using a 25 m (length), 0.53 mm (ID) and 0.50 μm (phase thickness) RTX-5 silica capillary column (Restek Corporation) and He (flow rate 50 cm / s) as carrier gas (split 1 : 20). Analysis conditions: injector: 220 °C, detector: 280 °C, oven temperature program: 70 °C (2 min) // 8 °C/min // 150 °C // 25 °C/min // 250 °C (5 min).

General procedure for the halogenation of isatin: a suspension of isatin (5 mmol), NXSac (5 mmol) and SiO₂ (5 g) in CH₂Cl₂ (25 mL) was stirred at r.t. for the time show in Table 1. The SiO₂ was filtered off, washed with MeOH and the organic solvents rotaevaporated. The residue was recrystallized from aqueous EtOH to produce pure 5-haloisatin. The physical data of the products are shown in Table 2.

2	v (cm ⁻¹ , KBr)	δ _H (ppm, CDCl ₃ / DMSO-d ₆)	δ _C (ppm, CDCl ₃ / DMSO-d ₆)	t _R (min)
X = C1	, , , ,		109.2 (CH), 113.8 (C), 120.0 (CH), 123.6 (C-Cl), 133.0 (CH), 144.6 (C), 154.4 [C(O)N], 178.9 [C(O)]	14.5
X = Br			113.8 (CH), 115.0 (C), 117.1 (C-Br), 126.9 (CH), 139.8 (CH), 149.2 (C), 158.3 [C(O)N], 182.9 [C(O)]	15.4
X = I		(m, 2H), 11.10 (broad s,	88.5 (C-I), 113.1 (CH), 119.5 (C), 133.0 (CH), 146.2 (CH), 152.1 (C), 160.2 [(C(O)N], 183.5 [C(O)]	_a

^a Decomposes during high-resolution gas chromatography analysis.

Table 2. Physical data of 5-haloisatins.

Acknowledgements: SPLS thanks CNPq for a fellowship. JFMS thanks CNPq and FUJB/UFRJ for financial support. We thank Prof. W. Bruce Kover and Joel Jones Jr for helpful discussions and Prof. Simon J. Garden for the samples of 5-haloisatins.

References

- 1. (a) A. McKillop and D.W. Young, Synthesis 401 (1979). (b) A. McKillop and D.W. Young, Synthesis 481 (1979). (c) J.H. Clark and D.J. Macquarrie, Chem. Soc. Rev. 25, 303 (1996). (d) M.C.S. de Mattos and A.M. Sanseverino, Synth. Commun. 30, 1975 (2000). (e) A.M. Sanseverino and M.C.S. de Mattos, J. Braz. Chem. Soc. 12, 685 (2001).
- 2. (a) I. Islam, D.D. Misra, R.N. Singh and J.P. Sharma, Talanta 31, 642 (1984). (b) A.G. Mistry, K. Smith and M.R. Bye, Tetrahedron Lett. 27, 1051 (1986). (c) K. Smith, D.M. James, A.G. Mistry, M.R. Bye and D.J. Faulkner, Tetrahedron 48, 7479 (1992). (d) M.R. Grimmett, Adv. Heterocycl. Chem. 57, 291 (1993). (e) P. Zhang, R. Liu and J.M. Cook, Tetrahedron Lett. 36, 3103 (1995). (f) R.N.P. Singh, A.K. Sinha and N. Sinha, Asian J. Chem. 10, 745 (1998). (g) F.Y. Miyake, K. Yakushijin and D.A. Horne, Org. Lett. 2, 2121 (2000). (h) S.M. Bonesi and R. Erra-Balsells, J. Heterocycl. Chem. 38, 77 (2001).
- 3. (a) M.G. Shvekhgeimer, Khim. Geterotsikl Soedin 291 (1996). (b) J.F.M. da Silva, S.J. Garden and A.C. Pinto, J. Braz. Chem. Soc. 12, 273 (2001).
- 4. (a) F.D. Popp, Adv. Heterocycl. Chem 18, 1 (1975). (b) M. Gopal, G. Srivastava, U.C. Pande and R.D. Tiwari, Mikrochim. Acta 11, 215 (1977). (c) S.J. Garden, J.C. Torres, S.C.D. Melo, A.S. Lima, A.C. Pinto and E.L.S. Lima, Tetrahedron Lett. 42, 2089 (2001).
- 5. J. Gasparic, T. Vontor, A. Lycka and D. Snobl, Collect. Czech. Chem. Commun. 55, 2963 (1990).
- 6. D. Dolenc, Synlett 544 (2000).
- 7. C. Berti and L. Greci, Synth. Commun. 11, 681 (1981).
- 8. S.N. Pandeya; D. Sriram, G. Nath and E. DeClercq, Eur. J. Pharm. Sci. 9, 25 (1999).
- 9. S.P.L. de Souza, J.F.M. da Silva and M.C.S. de Mattos, Synth. Commun. 33, 000 (2003).
- 10. A.M. Sanseverino, M.C.S. de Mattos and W.B. Kover, J. Chem. Res.(S) 346 (2000).

Received on October 15, 2002.